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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.003 Å R factor = 0.027 wR factor = 0.068 Data-to-parameter ratio = 21.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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A sterically crowded molecule: (*E*)-3,4-bis(bromomethyl)-2,2,5,5-tetramethylhex-3-ene

The title compound, $C_{12}H_{22}Br_2$, displays crystallographic inversion symmetry. Bond lengths C==C = 1.352 (4) Å and $Csp^2-Csp^3 = 1.574$ (3) Å are interpreted as symptoms of steric overcrowding. Molecules are linked into chains parallel to [101] by Br...Br interactions of 3.4275 (9) Å. Received 10 March 2006 Accepted 14 March 2006

Comment

The highly hindered diene 3,4-bismethylene-2,2,5,5-tetramethylhexane, (1), on treatment with bromine (in carbon tetrachloride at 273 K), provides predominantly the 1,4-addition product (E)-3,4-bis(bromomethyl)-2,2,5,5-tetramethylhex-3-ene, (2) (Backer, 1939). During our studies of highly hindered oligoenes (Hänel, 1996; Hopf *et al.*, 2001), we have repeated this experiment and obtained single crystals of (2) suitable for X-ray structural analysis.



The molecule of (2), which displays crystallographic inversion symmetry, is shown in Fig. 1. Because of this symmetry, IUPAC numbering cannot be used. Both the double bond C1=C1ⁱ and the single bond C1-C2 (symmetry code as in Table 1) may be regarded as significantly lengthened with respect to, for example, standard values of 1.331 Å for the central bond of the group $(C-)_2C=C(-C)_2$ and 1.522 Å for the central bond of $(C-)_3C-C=C$ (Allen *et al.*, 1987). The single bonds of the butyl group, in contrast, display normal lengths. The angle $C1^{i}-C1-C2$ is widened from the ideal 120° by *ca* 6° , at the expense of C6-C1-C2. These distortions may reasonably be attributed to steric pressure in the crowded system. Further evidence for this is provided by short intramolecular contacts $H3A \cdots H6B^{1}$ 1.89 and $H4B \cdots H6B^{1}$ 1.94 Å. The geometry at the double bond is essentially planar, with a maximum torsion angle of $-1.7 (4)^{\circ}$ about C1-C1¹. The configuration of the CH₂Br side chain is defined by the torsion angle $C2-C1-C6-Br = 78.2 (2)^{\circ}$.

The most striking feature of the crystal packing is the formation of zigzag chains of molecules, with overall direction parallel to [101], *via* Br···Br contacts of 3.4275 (9) Å (Fig. 2), symmetry operator $(1 - x, y, \frac{3}{2} - z)$. The C-Br···Br angles are equal by symmetry at 168.74 (6)°. This corresponds to a

 $D_x = 1.671 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 60 reflections

 $\theta = 10-11.5^{\circ}$ $\mu = 6.22 \text{ mm}^{-1}$ T = 173 (2) KBlock, colourless $0.5 \times 0.4 \times 0.4 \text{ mm}$

 $R_{\rm int}=0.035$

 $\theta_{\text{max}} = 27.5^{\circ}$ $h = -20 \rightarrow 20$

 $k = -9 \rightarrow 4$



Figure 1

A view of the title compound, with displacement ellipsoids drawn at the 50% probability level. [Symmetry code for unlabelled atoms: $\frac{1}{2} - x$, $\frac{1}{2} - y$, 1 - z.





A packing diagram of the title compound, viewed parallel to the b axis. Br \cdots Br contacts are indicated by dashed lines. H atoms have been omitted.

type I interaction according to the classification of Pedireddi *et al.* (1994).

Experimental

The diene, (1), was treated with bromine, as described by Backer (1939). The analytically pure adduct, (2), was obtained by preparative gas chromatography and characterized by NMR and MS analysis: ¹H NMR (20 MHz, CDCl₃, internal standard TMS, δ , p.p.m.): = 1.11 (*s*, 18H, *tert*-Bu), 4.1–4.9 (unresolved AB-*q*, 4H, CH₂). Single crystals were obtained by slow evaporation of a solution in petroleum ether (b.p. 333 K).

Crystal data

$C_{12}H_{22}Br_2$
$M_r = 326.12$
Monoclinic, C2/c
a = 15.658 (4) Å
b = 7.3521 (12) Å
c = 11.284 (2) Å
$\beta = 93.53 \ (2)^{\circ}$
V = 1296.6 (5) Å
Z = 4

Data collection

Stoe STAD14 diffractometer ω/θ scans Absorption correction: ψ scan (*XEMP*; Siemens, 1994) $T_{min} = 0.035$, $T_{max} = 0.083$ 2458 measured reflections 1487 independent reflections 1341 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $w = 1/[\sigma]$ $R[F^2 > 2\sigma(F^2)] = 0.027$ + 1.7 $wR(F^2) = 0.068$ whereS = 1.10 $(\Delta/\sigma)_{max}$ 1487 reflections $\Delta\rho_{max} =$ 70 parameters $\Delta\rho_{min} =$ H-atom parameters constrainedExtinction

0.083 $l = 0 \rightarrow 14$ ons3 standard reflectionsectionsfrequency: 60 min> $2\sigma(I)$ intensity decay: none

$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.0302P)^2 \\ &+ 1.7902P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} = 0.001 \\ \Delta\rho_{\text{max}} = 0.60 \text{ e } \text{\AA}^{-3} \\ \Delta\rho_{\text{min}} &= -0.57 \text{ e } \text{\AA}^{-3} \\ \text{Extinction correction: } SHELXL97 \\ \text{(Sheldrick, 1997)} \\ \text{Extinction coefficient: } 0.0023 (4) \end{split}$$

Table 1Selected geometric parameters (Å, °).

Br-C6	1.989 (2)	C2-C4	1.536 (3)
$C1-C1^{i}$	1.352 (4)	C2-C3	1.540 (3)
C1-C6	1.495 (3)	C2-C5	1.546 (3)
C1-C2	1.574 (3)		
$C1^{i}-C1-C6$	120.0 (2)	C6-C1-C2	113.99 (17)
$C1^{i}-C1-C2$	126.0 (2)		
C2-C1-C6-Br	78.20 (19)	$C6 - C1 - C1^i - C2^i$	-1.7 (4)
	1 1		

Symmetry code: (i) $-x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1.$

Methyl H atoms were located in difference syntheses, idealized $(C-H = 0.98 \text{ Å} \text{ and } H-C-H = 109.5^{\circ})$ and refined on the basis of rigid groups allowed to rotate but not tip. Methylene H atoms were placed in calculated opositions (C-H = 0.99 Å) and allowed to ride on their parent C atom. $U_{iso}(H)$ values were fixed at $1.2U_{eq}$ of the parent atom.

Data collection: *DIF4* (Stoe & Cie, 1992); cell refinement: *DIF4*; data reduction: *REDU4* (Stoe & Cie, 1992); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL97*.

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